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(54) Title: SHAPE MEMORY POLYURETHANE OR POLYURETHANE-UREA POLYMERS

(57) Abstract: The present invention relates to a shape memory polyurethane or polyurethane-urea polymer including a reaction product of: (A) (a) silicon-based macrodiol, silicon-based macrodiamine and/or polyether of the formula (I): A-[(CH<sub>2</sub>)m-O]n-(CH<sub>2</sub>)m-A', wherein A and A are endcapping groups; m is an integer of 6 or more; and n is an integer of 1 or greater; (b) a diisocyanate; and (c) a chain extender; or (B) (b) a diisocyanate: and (c) a chain extender, said polymer having a glass transition temperature which enables the polymer to be formed into a first shape at a temperature higher than the glass transition temperature and maintained in said first shape when the polymer is cooled to a temperature lower than the glass transition temperature, said polymer then being capable of resuming its original shape on heating to a temperature higher than the glass transition temperature. The present invention also relates to a shape memory composition which includes a blend of two or more of the shape memory polyurethane of polyurethane-urea polymers defined above or at least one shape memory polyurethane or polyurethane-urea polymer defined above in combination with another material. The present invention further relates to processes for preparing, materials having improved mechanical properties, clarity, processability, biostability and/or degradation resistance and devices or articles containing the shape memory polyurethane or polyurethane-urea polymer and/or composition defined above.

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#### SHAPE MEMORY POLYURETHANE OR POLYURETHANE-UREA POLYMERS

The present invention relates to polyurethane and polyurethane-urea polymers which have shape memory characteristics. The polymers respond to their shape memory when heated in a temperature range of about 20°C to about 100°C and are suitable for manufacturing articles, devices and implants requiring shape memory properties. The polymers are particularly useful in biomedical applications.

A shape memory polymer as a cast, moulded, foamed or extruded shape is capable of remembering a basic shape such as plane configuration and dead folds and taking on a second shape when the basic shape is modified1. The basic shape can be modified by changing the plane configuration and adding further folds, twists, kinks, bends and/or other three dimensional configurations at a temperature higher than the glass transition point  $(T_g)$  of the polymer, but lower than the moulding temperature. The modified shape is typically set when the polymer is cooled in the modified state to a temperature lower than the glass transition temperature. The method of utilising the shape memory is by heating the modified shape to a temperature higher than the glass transition temperature thereby restoring the original shape. Polymers with such characteristics combined with biostability would find many applications in the fabrication of various medical devices. The device shape can be optimised depending on the location site, for example, the shape could be modified by coiling or collapsing and subsequent cooling to a temperature below the glass transition temperature to freeze the modified shape. Thermally triggered shape memory could then occur thereby returning the device to its original shape to enable fixing or anchoring to the location site. Medical devices which would benefit from such shape memory characteristics include bone suture anchors, vascular, esophageal and bilial stents and cochlear implantations.

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Segmented copolymers such as thermoplastic polyurethanes usually exhibit shape memory characteristics if formulated such that the glass transition temperature of one segment falls within a useful temperature range of about 25°C to about 60°C. Such polyurethanes are generally prepared from polyester or polyether macrodiols, aromatic diisocyanates and chain extenders<sup>1,2,3</sup>.

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The shape memory polyurethane compositions disclosed in United States Patent Nos. 5,049,591 and 5,139,832 are formulated with conventional reagents used in the art of polyurethane manufacture and hence are prone to degradation, particularly under the oxidative and hydrolytic conditions present in biological environments.

implant applications is expected to be very poor since commercial polyurethanes such as Estane are based on degradation-prone<sup>4,5</sup> polytetramethylene oxide (PTMO), 4,4'-diphenylmethane diisocyanate and 1,4 - butanediol. Similarly, polycarbonate macrodiol based shape memory polyurethanes are expected to have very poor hydrolytic resistance and be unsuitable for long term medical implants<sup>6</sup>. These commercial polyurethanes often also contain small amounts of low molecular weight residues and additives that leach out of the polyurethane and cause undesirable biological responses.

United States Patent No. 5814705 discloses shape memory compositions based on blends of commercial polyurethanes such as Estane with other block copolymers. The compatibility of the component polymers may not be sufficient to have a homogeneous shape memory polymer composition. Such compositions, particularly in long term use, may lead to poor performance due to a phase separation of the component polymers.

A range of biostable polyurethanes are disclosed in International Patent Publication Nos. W098/13405 and W099/03863 and United States Patent No. 5,393,858. We have found that by proper choice of components and the relative

amounts of the hard and soft segments that biostable polyurethanes can be formulated to have one glass transition temperature in a temperature range of about 20°C to about 100°C. Such polyurethanes therefore possess both the properties of biostability, compatibility and shape memory which enable them to be used in the manufacture of medical articles, devices and implants.

According to the present invention there is provided a shape memory polyurethane or polyurethane-urea polymer including a reaction product of:

(A) (a) silicon-based macrodiol, silicon-based macrodiamine and/or polyethers of the formula (I):

$$A - \left[ (CH_2)_m - O \right]_n - (CH_2)_m - A'$$

wherein

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A and A' are endcapping groups; m is an integer of 6 or more; and n is an integer of 1 or greater;

- (b) a diisocyanate; and
- (c) a chain extender; or
- (B) (b) a diisocyanate; and
  - (c) a chain extender,

said polymer having a glass transition temperature which enables the polymer to be formed into a first shape at a temperature higher than the glass transition temperature and maintained in said first shape when the polymer is cooled to a temperature lower than the glass transition temperature, said polymer then being capable of resuming its original shape on heating to a temperature higher than the glass transition temperature.

The term "endcapping group" is used herein in its broadest sense and includes reactive functional groups or groups containing reactive functional groups. Suitable examples of reactive functional groups are alcohols, carboxylic acids, aldehydes, ketones, esters, acid halides,

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acid anhydrides, amines, imines, thio, thioesters, sulphonic acid and expoxides. Preferably the reactive functional group is an alcohol or an amine, more preferably an alcohol.

Further according to the present invention there is provided a shape memory polyurethane or polyurethane-urea composition which includes at least one of the shape memory polyurethane or polyurethane-urea polymers defined above and optionally another material.

Component (a) is preferably a combination of at least two macrodiols, at least two macrodiamines or at least one macrodiol and at least one macrodiamine. Shape memory polymers in which component (a) is a combination of at least two macrodiols or at least one macrodiol and at least one macrodiamine preferably have greater than about 50% silicon-based macrodiol, in particular greater than about 70% as such polymers possess good biostability. A suitable molecular weight range of component (a) is about 300 to about 2000, more preferably about 300 to about 700.

The silicon-based macrodiol or macrodiamine may be a polysilane, polysiloxane, amino-terminated polysiloxane or a silicon-based polycarbonate.

The polysiloxane or amino-terminated polysiloxane may be represented by the formula (II):

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$$A-R_{5}-\begin{array}{c} R_{1} \\ \downarrow \\ R_{3} \end{array} \begin{bmatrix} R_{2} \\ R_{7}-\begin{array}{c} \downarrow \\ \downarrow \\ R_{4} \end{bmatrix} \\ R_{6}-A'$$
(II

wherein

A and A' are as defined above;

R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub>, R<sub>5</sub> and R<sub>6</sub> are the same or different and selected from hydrogen or an optionally substituted straight chain, branched or cyclic, saturated or unsaturated hydrocarbon radical;

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R<sub>7</sub> is a divalent linking group or an optionally substituted straight chain, branched or cyclic, saturated or unsaturated hydrocarbon radical; and

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p is an integer of 1 or greater.

5 The hydrocarbon radical for substituents R, R1, R2, R3 and R4 may include alkyl, alkenyl, alkynyl, aryl or heterocyclyl radicals. It will be appreciated that the equivalent radicals may be used for substituents  $R_5$ ,  $R_6$  and R7 except that the reference to alkyl, alkenyl and alkynyl should be to alkylene, alkenylene and alkynylene, 10 respectively. In order to avoid repetition, only detailed definitions of alkyl, alkenyl and alkynyl are provided

hereinafter. The term "alkyl" denotes straight chain, branched 15 or mono- or poly-cyclic alkyl, preferably C1-12 alkyl or cycloalkyl. Examples of straight chain and branched alkyl include methyl, ethyl, propyl, isopropyl, butyl, isobutyl, sec-butyl, amyl, isoamyl, sec-amyl, 1,2-dimethylpropyl, 1,1-dimethylpropyl, pentyl, hexyl, 4-methylpentyl, 20 1-methylpentyl, 2-methylpentyl, 3-methylpentyl, 1,1-dimethylbutyl, 2,2-dimethylbutyl, 3,3-dimethylbutyl, 1,2-dimethylbutyl, 1,3-dimethylbutyl, 1,2,2trimethylpropyl, 1,1,2-trimethylpropyl, heptyl, 5-methylhexyl, 1-methylhexyl, 2,2-dimethylpentyl, 3,3-dimethylpentyl, 4,4-dimethylpentyl, 1,2-dimethylpentyl, 25 1,3-dimethylpentyl, 1,4-dimethylpentyl, 1,2,3-trimethylbutyl, 1,1,2-trimethylbutyl, 1,1,3-trimethylbutyl, octyl, 6-methylheptyl, 1methylheptyl, 1,1,3,3-tetramethylbutyl, nonyl, 1-, 2-, 3-, 4-, 5-, 6- or 7-methyloctyl, 1-, 2-, 3-, 4- or 5ethylheptyl, 1-, 2- or 3-propylhexyl, decyl, 1-, 2-, 3-,

4-, 5-, 6-, 7- and 8-methylnonyl, 1-, 2-, 3-, 4-, 5- or 6-ethyloctyl, 1-, 2-, 3- or 4-propylheptyl, undecyl, 1-, 2-, 3-, 4-, 5-, 6-, 7-, 8- or 9-methyldecyl, 1-, 2-, 3-, 4-, 5-, 6- or 7-ethylnonyl, 1-, 2-, 3-, 4- or

5-propyloctyl, 1-, 2- or 3-butylheptyl, 1-pentylhexyl, dodecyl, 1-, 2-, 3-, 4-, 5-, 6-, 7-, 8-, 9- or

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10-methylundecyl, 1-, 2-, 3-, 4-, 5-, 6-, 7- or 8-ethyldecyl, 1-, 2-, 3-, 4-, 5- or 6-propylnonyl, 1-, 2-, 3- or 4-butyloctyl, 1,2-pentylheptyl and the like. Examples of cyclic alkyl include cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cycloheptyl, cycloctyl, cyclononyl, cyclodecyl and the like.

The term "alkenyl" denotes groups formed from straight chain, branched or mono- or poly-cyclic alkenes including ethylenically mono- or poly-unsaturated alkyl or 10 cycloalkyl groups as defined above, preferably C2-12 alkenyl. Examples of alkenyl include vinyl, allyl, 1-methylvinyl, butenyl, iso-butenyl, 3-methyl-2-butenyl, 1-pentenyl, cyclopentenyl, 1-methyl-cyclopentenyl, 1-hexenyl, 3-hexenyl, cyclohexenyl, 1-heptenyl, 3-heptenyl, 15 1-octenyl, cyclooctenyl, 1-nonenyl, 2-nonenyl, 3-nonenyl, 1-decenyl, 3-decenyl, 1,3-butadienyl, 1,4-pentadienyl, 1,3cyclopentadienyl, 1,3-hexadienyl, 1,4-hexadienyl, 1,3-cyclohexadienyl, 1,4-cyclohexadienyl, 1,3-cycloheptadienyl, 1,3,5-cycloheptatrienyl, 20 1,3,5,7-(cycloocta-tetraenyl) and the like.

The term "alkynyl" denotes groups formed from straight chain, branched, or mono- or poly-cyclic alkynes. Examples of alkynyl include ethynyl, 1-propynyl, 1- and 2-butynyl, 2-methyl-2-propynyl, 2-pentynyl, 3-pentynyl, 4-pentynyl, 2-hexynyl, 3-hexynyl, 4-hexynyl, 5-hexynyl, 10-undecynyl, 4-ethyl-1-octyn-3-yl, 7-dodecynyl, 9-dodecynyl, 10-dodecynyl, 3-methyl-1-dodecyn-3-yl, 2-tridecynyl, 11-tridecynyl, 3-tetradecynyl, 7-hexadecynyl, 3-octadecynyl and the like.

The term "aryl" denotes single, polynuclear, conjugated and fused residues of aromatic hydrocarbons.

Examples of aryl include phenyl, biphenyl, terphenyl, quaterphenyl, phenoxyphenyl, naphthyl, tetrahydronaphthyl, anthracenyl, dihydroanthracenyl, benzanthracenyl, dibenzanthracenyl, phenanthrenyl and the like.

The term "heterocyclyl" denotes mono- or poly-cyclic heterocyclyl groups containing at least one

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heteroatom selected from nitrogen, sulphur and oxygen. Suitable heterocyclyl groups include N-containing heterocyclic groups, such as, unsaturated 3 to 6 membered heteromonocyclic groups containing 1 to 4 nitrogen atoms, for example, pyrrolyl, pyrrolinyl, imidazolyl, pyrazolyl, pyridyl, pyrimidinyl, pyrazinyl, pyridazinyl, triazolyl or tetrazolyl; saturated 3 to 6 membered heteromonocyclic groups containing 1 to 4 nitrogen atoms, such as pyrrolidinyl, imidazolidinyl, piperidino or piperazinyl; 10 unsaturated condensed heterocyclic groups containing 1 to 5 nitrogen atoms, such as, indolyl, isoindolyl, indolizinyl, benzimidazolyl, quinolyl, isoquinolyl, indazolyl, benzotriazolyl or tetrazolopyridazinyl; unsaturated 3 to 6-membered heteromonocyclic group containing an oxygen 15 atom, such as, pyranyl or furyl; unsaturated 3 to 6-membered heteromonocyclic group containing 1 to 2 sulphur atoms, such as, thienyl; unsaturated 3 to 6-membered heteromonocyclic group containing 1 to 2 oxygen atoms and 1 to 3 nitrogen atoms, such as, oxazolyl, isoazolyl or oxadiazolyl; saturated 3 to 6-membered heteromonocyclic 20 group containing 1 to 2 oxygen atoms and 1 to 3 nitrogen atoms, such as, morpholinyl; unsaturated condensed heterocyclic group containing 1 to 2 oxygen atoms and 1 to 3 nitrogen atoms, such as, benzoxazolyl or benzoxadiazolyl; 25 unsaturated 3 to 6-membered heteromonocyclic group containing 1 to 2 sulphur atoms and 1 to 3 nitrogen atoms, such as thiazolyl or thiadiazolyl; saturated 3 to 6-membered heteromonocyclic group containing 1 to 2 sulphur atoms and 1 to 3 nitrogen atoms, such as, thiadiazolyl; and 30 unsaturated condensed heterocyclic group containing 1 to 2 sulphur atoms and 1 to 3 nitrogen atoms, such as benzothiazolyl or benzothiadiazolyl.

In this specification, "optionally substituted" means that a group may or may not be further substituted with one or more groups selected from oxygen, nitrogen, sulphur, alkyl, alkenyl, alkynyl, aryl, halo, haloalkyl, haloalkenyl, haloalkynyl, haloaryl, hydroxy, alkoxy,

alkenyloxy, alkynyloxy, aryloxy, carboxy, benzyloxy, haloalkoxy, haloalkenyloxy, haloalkynyloxy, haloaryloxy, nitro, nitroalkyl, nitroalkenyl, nitroalkynyl, nitroaryl, nitroheterocyclyl, azido, amino, alkylamino, alkenylamino, alkynylamino, arylamino, benzylamino, acyl, alkenylacyl, alkynylacyl, arylacyl, acylamino, acyloxy, aldehydo, alkylsulphonyl, arylsulphonyl, alkylsulphonylamino, arylsulphonylamino, alkylsulphonyloxy, arylsulphonyloxy, heterocyclyl, heterocycloxy, heterocyclylamino, haloheterocyclyl, alkylsulphenyl, arylsulphenyl, 10 carboalkoxy, carboaryloxy, mercapto, alkylthio, arylthio, acylthio and the like.

Suitable divalent linking groups for R7 include O, S and NR wherein R is hydrogen or an optionally substituted straight chain, branched or cyclic, saturated or unsaturated hydrocarbon radical.

Preferred polysiloxanes are polysiloxane macrodiols which are polymers of the formula (II) wherein R and R' are hydroxy and include those represented by the formula (III):

HO-
$$R_5$$
- $S_1$ - $\begin{bmatrix} R_2 \\ O-S_1 \\ R_3 \end{bmatrix}$   $R_6$ -OH

wherein

 $R_1$  to  $R_6$  and p are as defined in formula (II)

30 above.

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A preferred polysiloxane is PDMS which is a compound of formula (III) wherein  $R_1$  to  $R_4$  are methyl and  $R_5$ and  $R_6$  are as defined above. Preferably  $R_5$  and  $R_6$  are the same or different and selected from propylene, butylene, pentylene, hexylene, ethoxypropyl (-CH2CH2OCH2CH2CH2-), propoxypropyl and butoxypropyl.

The polysiloxane macrodiols may be obtained as

commercially available products such as X-22-160AS from Shin Etsu in Japan or prepared according to known procedures. The preferred molecular weight range of the polysiloxane macrodiol is about 200 to about 6000, more preferably about 500 to about 2000.

Other preferred polysiloxanes are polysiloxane macrodiamines which are polymers of the formula (II) wherein A is NH2, such as, for example, amino-terminated PDMS.

10 Suitable silicon-based polycarbonates include those described in our International Patent Publication No. WO98/54242, the entire content of which is incorporated herein by reference.

A preferred silicon-based polycarbonate has the 15 formula (IV):

$$\begin{array}{c|c}
R_{2} \\
R_{7} - S_{1} \\
R_{4}
\end{array} - R_{6} - \begin{bmatrix}
O \\
O - R_{8}
\end{bmatrix}_{y} - \begin{bmatrix}
O \\
O - C - O - R_{9}
\end{bmatrix}_{z} - \begin{bmatrix}
O \\
R_{1} \\
- O - C - O - R_{5} - S_{1} - \begin{bmatrix}
R_{2} \\
R_{7} - S_{1} \\
R_{3}
\end{bmatrix}_{m} - R_{6} - A'$$
(IV)

wherein

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30  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$ ,  $R_5$ ,  $R_6$  and  $R_7$  are as defined in formula (I) above;

R<sub>8</sub> and R<sub>9</sub> are same or different and selected from hydrogen or an optionally substituted straight chain, branched or cyclic, saturated or unsaturated hydrocarbon radical;

> A and A' are as defined in formula (I) above; m, y and z are integers of 0 or more; and

x is an integer of 0 or more.

Preferably z is an integer of 0 to about 50 and x is an integer of 1 to about 50. Suitable values for m include 0 to about 20, more preferably 0 to about 10. Preferred values for y are 0 to about 10, more preferably 0 to about 2.

A preferred polycarbonate is a compound of the formula (IV) wherein the endcapping group is a hydroxy which is a polycarbonate macrodiol of the formula (IVa):

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$$\begin{array}{c|c} R_1 & R_2 \\ H-O-R_5-Si- & R_3 & R_4 \\ \end{array} \\ \begin{array}{c|c} R_2 \\ R_7-Si \\ R_4 \\ \end{array} \\ \begin{array}{c|c} R_6-O-C \\ \end{array} \\ \begin{array}{c|c} O-R_5- & Si- \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_2 \\ R_7-Si \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_2 \\ R_7-Si \\ R_4 \\ \end{array} \\ \begin{array}{c|c} R_2 \\ R_6-O-C \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_5-Si- \\ R_7-Si- \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_7-Si- \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_2 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_4 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_5-O-C \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_2 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_4 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_2 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_2 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_2 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_2 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_2 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_2 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_2 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_3 \\ \end{array} \\ \begin{array}{c|c} R_1 \\ R_2 \\ R_3 \\ \end{array}$$

$$20 \begin{bmatrix} R_{2} \\ R_{7} - Si \\ R_{4} \end{bmatrix}_{m} - R_{6} - \begin{bmatrix} O \\ O - C - O - R_{9} \end{bmatrix}_{y} - \begin{bmatrix} O \\ O - C - O - R_{9} \end{bmatrix}_{z} - O - C - O - R_{5} - Si - \begin{bmatrix} R_{2} \\ R_{7} - Si \\ R_{3} \end{bmatrix}_{m} - R_{6} - O - H$$
(IVa)

wherein

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 $R_1$  to  $R_9$ , m, y, x and z are as defined in formula (IV) above.

Particularly preferred polycarbonate macrodiols are compounds of the formula (IVa) wherein  $R_1$ ,  $R_2$ ,  $R_3$  and  $R_4$  are methyl,  $R_8$  is ethyl,  $R_9$  is hexyl,  $R_5$  and  $R_6$  are propyl or  $R_4$  butyl and  $R_7$  is 0 or -CH<sub>2</sub>-CH<sub>2</sub>-,more preferably  $R_5$  and  $R_6$  are propyl when  $R_7$  is 0 and  $R_5$  and  $R_6$  are butyl when  $R_7$  is -CH<sub>2</sub>-CH<sub>2</sub>-. The preferred molecular weight range of the polycarbonate macrodiol is about 400 to about 5000, more preferably about 400 to about 2000.

Suitable polyethers include polyether macrodiols represented by the formula (V):

$$HO-[(CH_2)_m-O]_n-H$$

5 wherein

m is as defined in formula (I) above, preferably 6 to 18; and

n is as defined in formula (I) above, preferably 1 to 50.

Polyether macrodiols of formula (V) wherein m is 6 or higher such as poly(hexamethyleneoxide) (PHMO), poly(heptamethyleneoxide), poly(octamethylene oxide) (POMO) and poly(decamethylene oxide) (PDMO) are preferred over the conventional PTMO. PHMO and PDMO are particularly preferred due to their relatively high glass transition temperatures.

The polyether macrodiols may be prepared by the procedure described by Gunatillake et al<sup>6</sup>. The preferred molecular weight range of the polyether macrodiol is about 300 to about 2000, more preferably about 300 to about 700.

In a particularly preferred embodiment, component

(a) is a combination of PDMS or amino-terminated PDMS with
another polymer falling within the scope of component (a),
for example, a polyether of the formula (I) such as PHMO or
a silicon-based polycarbonate such as siloxy carbonate.

The diisocyanates may be aliphatic or aromatic diisocyanates such as, for example 4,4'-diphenylmethane diisocyanate (MDI), methylene biscyclohexyl diisocyanate (H<sub>12</sub>MDI), p-phenylene diisocyanate (p-PDI), trans-cyclohexane-1,4-diisocyanate (CHDI),

1,6-diisocyanatohexane (DICH), 1,5-diisocyanatonaphthalene (NDI), para-tetramethylxylenediisocyanate (p-TMXDI), meta-tetramethylxylene diisocyanate (m-TMXDI), 2,4-toluene diisocyanate (2,4-TDI) isomers or mixtures thereof or isophorone diisocyanate (IPDI). MDI is particularly preferred.

The chain extender may be selected from diol or diamine chain extenders. Examples of diol chain extenders

include 1,4-butanediol, 1,6-hexanediol, 1,8-octanediol,
1,9-nonanediol, 1,10-decanediol, 1,12-dodecanediol,1,4cyclohexanediol, 1,4-cyclohexanedimethanol, p-xyleneglycol,
1,3-bis(4-hydroxybutyl) tetramethyldisiloxane,

1,3-bis(6-hydroxyethoxypropyl)tetramethyldisiloxane and 1,4-bis(2-hydroxyethoxy)benzene. Suitable diamine chain extenders include 1,2-ethylenediamine,

1,3-propanediamine, 1,4-butanediamine,

1,3-bis(3-aminopropyl)tetramethyldisiloxane,

10 1,3-bis(4-aminobutyl)tetramethyldisiloxane and 1,6-hexanediamine.

The chain extender may also be a silicon-containing chain extender of the type described in our International Patent Publication No. W099/03863, the entire contents of which are incorporated herein by reference. Such chain extenders include a silicon-containing diol of the formula (VI):

$$HO-R_5-Si-\begin{bmatrix}R_1\\R_7-Si\\R_3\end{bmatrix}-R_6-OH$$

$$R_3\begin{bmatrix}R_2\\R_7-Si\\R_4\end{bmatrix}$$

wherein

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 $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$ ,  $R_5$ ,  $R_6$  and  $R_7$  are as defined in formula (II) above; and

q is 0 or greater, preferably 2 or less.

Component (a) of the polymer generally forms the soft segment of the polyurethane or polyurethane-urea and provides the low glass transition temperature. The high glass transition temperature is provided by the hard segment components (b) and (c).

Preferably, the amount of hard segment in the polymer of the present invention is about 30 to about 100 wt%, more preferably about 50 to about 80 wt%, most preferably about 60 to about 70 wt%. However, it will be appreciated that this amount is dependent on the type of

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soft segment polymer used, in particular the molecular weight of this polymer. For example, when the molecular weight of the soft segment polymer is about 500, then a 55 to 60 wt% hard segment is preferred. If the molecular weight of the soft segment polymer is about 1000, then a 45 to 55 wt% hard segment is preferred.

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For most applications, it is preferred that the shore hardness of the polymer below the glass transition temperature is in the range of about 82D to about 50D, while the hardness above the glass transition temperature is in the range of about 20D to about 30D. The glass transition temperatures of the polymers and compositions of the present invention are generally in the range of about 20°C to about 100°C, preferably about 20 to about 60°C. However, in some applications such as biotechnological applications, it may be advantageous for the glass transition temperature to be sub ambient i.e., below about 20°C.

compositions of the present invention may include a blend of two or more of the shape memory polyurethane or polyurethane-urea polymers defined above or at least one shape memory polyurethane or polyurethane-urea polymer defined above in combination with another material. The other material will preferably be of any suitable known type which does not substantially effect the shape memory and/or biostability properties of the polymers of the present invention and may include polymeric and non-polymeric materials.

Examples of polymeric materials include conventional polyurethanes such as PELETHANE<sup>TM</sup>, ESTANE<sup>TM</sup>, CARBOTHANE<sup>TM</sup>, CORETHANE<sup>TM</sup> and CHRONOFLEX<sup>TM</sup>; shape memory polyurethanes such as those disclosed in United States Patent Nos. 5,145,935 and 5,135,786 and available from Mitsubishi Heavy Industries Ltd (distributed by Memry Corporation in the United States of America; polyolefins such as polyethylene, polypropylene, ethylene propylene

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copolymers, metallocene polymers, ethylene vinylacetate copolymers and polyvinyl chloride; polyamides; and liquid crystalline polymers such as those available from Eastman Kodak (XG7), Mitsubishi Chemical Industry (Novaculates) and Idemitsu Petrochemical Industry (Idemitsu LCP and Unitika (Lodrum LC). Such polymeric materials generally blend well with the shape memory polymers of the present invention which usually contain high levels of polysiloxane segments.

Each of the polymers forming the shape memory 10 composition preferably have different glass transition temperatures and/or different amounts of hard segment component. Suitable compositions may include a first polymer with a low glass transition temperature, preferably below about ambient temperature and a second polymer with a glass transition temperature above the ambient temperature, 15 more preferably above about 50°C. The two polymers can be blended in proportions such that the final blend will have a glass transition temperature in the preferred range of about 20°C to about 60°C. Generally the glass transition 20 temperature of the composition is intermediate to those of the two polymers.

Alternatively, the composition may include a first polymer having a high percentage of hard segment component, for example, above about 70 wt%, more preferably above about 90%. Particularly preferred examples of such polymers are the non-elastomeric polyurethane or polyurethane-urea polymers disclosed in International Patent Application No. PCT/AU99/00236. This first polymer can be blended with a second polymer having a lower percentage of hard segment, for example, about 30 to about 60 wt%, more preferably about 40 to about 50 wt%. of suitable polymeric blends include a combination of an elastomeric and a non-elastomeric polyurethane or polyurethane-urea polymer. The term "non-elastomeric" refers to polyurethanes having a % elongation of up to about 200% generally up to about 100%. This technique allows a composition having a softening temperature

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appropriate for the application to be prepared.

The shape memory polymers and compositions of the present invention may be prepared by any technique familiar to those skilled in the manufacture of polyurethanes. These include one or two-step bulk or solution polymerisation procedures. The polymerisation can be carried out in conventional apparatus or within the confines of a reactive extruder continuous injection moulding or mixing machines.

In a one-step bulk polymerisation procedure the appropriate amount of component (a) is mixed with the chain extender first at temperatures in the range of about 45 to about 100°C, more preferably about 60 to about 80°C. If desired a catalyst such as stanneous octoate or dibutyltin dilaurate at a level of about 0.001 to about 0.5 wt % based on the weight of the total ingredients may be added to the initial mixture. Molten diisocyanate is then added and mixed thoroughly to give a homogeneous polymer liquid and cured by pouring the liquid polymer into Teflon-coated trays and heating in an oven to about 100°C.

The shape memory polymers can also be prepared by a two-step method where a prepolymer is prepared by reacting component (a) with a diisocyanate. The prepolymer is then reacted with a suitable chain extender.

The polymers and compositions of the present invention are particularly useful in preparing materials having good mechanical properties, more specifically biomaterials as a consequence of their biostability or improved resistance to degradation and their shape memory properties.

According to another aspect of the present invention there is provided a material having improved mechanical properties, clarity, processability, biostability and/or degradation resistance including the polymer or composition defined above.

The present invention also provides use of the polymer or composition defined above as a material having improved mechanical properties, clarity, processability,

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biostability and/or degradation resistance.

The present invention further provides the polymer or composition defined above when used as a material having improved mechanical properties, clarity, processability, biostability and/or degradation resistance.

The mechanical properties which are improved include tensile strength, tear strength, flex fatigue resistance, abrasion resistance, Durometer hardness, flexural modulus and related measures of flexibility or elasticity.

The improved resistance to degradation includes resistance to free radical, oxidative, enzymatic and/or hydrolytic processes and to degradation when implanted as a biomaterial.

15 The improved processability includes ease of processing by casting such as solvent casting and by thermal means such as extrusion and injection molding, for example, low tackiness after extrusion and relative freedom from gels.

The term "biostability" is used herein in its broadest sense and refers to a stability when in contact with cells and/or bodily fluids of living animals or humans.

There is also provided a biostable material which includes the polymer or composition defined above.

There is further provided a degradation resistant material which includes the polymer or composition defined above.

The polymer or composition of the present invention should also have a good compatibility and stability in biological environments, particularly when implanted in vivo for extended periods of time.

According to another aspect of the present invention there is provided an in vivo degradation resistant or biostable material which includes the polymer or composition defined above.

The polymer or composition may also be used as a

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biomaterial. The term "biomaterial" is used herein in its broadest sense and refers to a material which is used in situations where it comes into contact with the cells and/or bodily fluids of living animals or humans.

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The polymer or composition is therefore useful in manufacturing medical devices, articles or implants.

Thus, the present invention still further provides medical devices, articles or implants which are composed wholly or partly of the polymer or composition defined above.

The medical devices, articles or implants may include catheters; stylets; bone suture anchors; vascular, oesophageal and bilial stents; cochlear implants; reconstructive facial surgery; controlled drug release devices; components in key hole surgery; biosensors; membranes for cell encapsulations; medical guidewires; medical guidepins; cannularizations; pacemakers, defibrillators and neurostimulators and their respective electrode leads; ventricular assist devices; orthopaedic joints or parts thereof including spinal discs and small joints; cranioplasty plates; intraoccular lenses; urological stents and other urological devices; stent/graft devices; device joining/extending/repair sleeves; heart valves; vein grafts; vascular access ports; vascular 25 shunts; blood purification devices; casts for broken limbs; vein valve, angioplasty, electrophysiology and cardiac output catheters; and tools and accessories for insertion of medical devices, infusion and flow control devices.

As the polymers and compositions of the present invention may be designed so that they are rigid at ambient 30 temperature but soften around the body temperature they have many applications in the construction of medical articles, devices and implants. For example, intravenous catheters made from such materials could be inserted 35 initially in the vein due to the high flexural modulus of the material, but would then soften once inside the blood vessel. Furthermore, catheters may be modified to a

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predetermined shape for ease of directing to a target area or modified in such a way to have sections with different softening temperatures, for ease of guidance of the device to a specific location.

It will be appreciated that polymers and compositions having properties optimised for use in the construction of various medical devices, articles or implants and possessing shape memory characteristics will also have other non-medical applications. Such applications may include toys and toy components, shape memory films, pipe couplings, electrical connectors, zero-insertion force connectors, Robotics, Aerospace actuators, dynamic displays, flow control devices, sporting goods and components thereof, body-conforming devices, temperature control devices, safety release devices and heat shrink insulation.

Thus, the present invention extends to the use of the polymer or composition defined above in the manufacture of devices or articles.

The present invention also provides devices or articles which are composed wholly or partly of the polymer or composition as defined above.

The invention will now be described with reference to the following non-limiting examples.

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## EXAMPLE 1

Poly(hexamethylene oxide) (PHMO) (MW 489.7) was prepared according to a method described by Gunatillake et al<sup>7</sup> and United States Patent No. 5,403,912 and dried at 130°C under vacuum for 4 h. A shape memory polyurethane composition from PHMO was prepared according to a one-step bulk polymerisation as described below.

PHMO (35.00 g) and 1,4-butanediol (BDO) (12.06 g) were weighed in to a 500 mL polypropylene beaker and the contents warmed to 70°C. Molten MDI (52.93 g) was weighed into a 100 mL, wet-tared polypropylene beaker and added to the PHMO/BDO mixture quickly with stirring. The mixture was

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stirred for about 30 sec and the contents poured onto a Teflon-coated metal pan. The polyurethane was cured at 100°C for 4 h under nitrogen. The resulting polyurethane was clear and transparent. The specimens for various tests were prepared by compression molding at a temperature of 200°C and injection moulding.

Dynamical Mechanical Thermal Analysis, DMTA (Rheometrics MkIIIe) was performed on 40 mm x 10 mm x 1 mm compression moulded samples in single cantilever bending mode at Htz over a temperature range of 30°C to 90°C at a ramp rate of 2°C/min. The onset of the change in the bending modulus was at 37°C (1100±50 MPa bending mod) and the endset of change in the bending modulus was 56°C (50±20 MPa bending modulus).

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The shape memory characteristics of the polyurethane composition were demonstrated as follows. An injection moulded flat, 2.5 mm thick plaque of the polyurethane and a compression moulded flat thin film (0.1 mm thick) were folded 180° at 55°C and cooled to 20°C so that the plaque and thin film were locked into a 180° folded configuration. The folded plaque and the thin film were stored for 72 hours without any configurational change and then subsequently heated in water to 55°C at which point the folded thin film very quickly (<1 sec) returned to its original flat configuration and the thicker plaque returned also to its original flat configuration but more slowly (ca. 20 secs).

A reverse experiment was also performed whereby permanent 180° folds were placed in the samples by compressing between flat plates heated to 150°C. The thick and thin samples were then heated to 55°C and the 180° fold undone to 0°, this unfolding being locked in by cooling to 20°C. The samples were stored at ambient temperature for 72 hours in the modified (unfolded) shape with no observable configuration changes. The samples were subsequently heated in water at 55°C causing the original 180° fold to reform in similar times to those observed in the previous

experiment.

#### EXAMPLE 2

A polyurethane based on PHMO with a molecular weight of 398.0 was prepared using a procedure similar to 5 that described in Example 1. PHMO (32.00 g) and 1,4butanediol (12.67 g) was weighed into a 500 mL polypropylene beaker and the contents warmed to 70°C. Molten MDI (55.33 g) was weighed into a 100 mL wet-tared polypropylene beaker and added to the PHMO quickly with 10 stirring. The mixture was stirred for about 30 sec and the contents poured onto a Teflon-coated metal pan. The polyurethane was cured at 100°C for 4 h under nitrogen. The resulting polyurethane was clear and transparent. specimens for various tests were prepared by compression 15 moulding at a temperature of 200°C and injection moulding.

The onset of the change in the bending modulus was at  $46^{\circ}$ C ( $1050\pm50$  MPa bending mod) and the endset of change in the bending modulus was  $60^{\circ}$ C ( $50\pm20$  MPa bending mod) as determined by DMTA analysis. The shape memory characteristics of the composition was similar to that of the composition of Example 1.

#### EXAMPLE 3

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This example illustrates the preparation of shape memory polyurethane compositions with desired glass transition temperatures in the 20°C to 100°C range by solvent blending of two polyurethane compositions, one with a low flexural modulus (approximately in the range of about 15 to about 100 MPa range) and the other with a high flexural modulus (> 500MPa).

The low modulus polyurethane composition was prepared by reacting bis(6-hydroxyethoxypropyl) polydimethylsiloxane (48.00 g, MW 940.3), poly(hexamethylene oxide) (12.00 g, MW 700.2), 1,4-butanediol (5.80 g) and MDI (34.19 g) according to a one-step polymerisation procedure. The flexural modulus of

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the polyurethane was 30 MPa.

The high modulus polyurethane composition was prepared by reacting 1,4-cyclohexanedimethanol (25.27g), 1,3bis(4-hydroxybutyl)-1,1,3,3-tetramethyldisiloxane (16.27g) and MDI (58.46g) according to a one-step bulk polymerisation. The flexural modulus of the polyurethane was 1770 MPa.

Differential scanning calorimetry (at a ramp rate of 10°C/min) demonstrated the presence of glass transition change onset at 91.2°C and an endset at 106.7°C with a Cp of 0.28J.g<sup>-1</sup>.°C<sup>-1</sup>. This high modulus composition exhibited shape memory characteristics. A compression moulded thin plaque (0.1 mm) was folded at 110°C and immediately cooled to ambient temperature to preserve the fold. It was subsequently heated to 110°C resulting in a reversal of the shape to the original.

The high modulus and low modulus polyurethanes were blended by mixing 7.5 g and 2.5 g, respectively and dissolving the blend in N,N-dimethylformamide to give a 20 wt% solution. A thin film of the blend was prepared by solvent casting. The polymer solution was poured onto a Petrie Dish to form a 5 mm thick layer and the solvent evaporated in a nitrogen circulating oven over a period of 48 h. DSC analysis of the dried film showed a glass transition onset temperature of 45.6°C and an end set at 49.5°C.

A thin film (0.3 mm) of the blend was folded by 180° by heating to a temperature above 50°C and the folded shape fixed by cooling to room temperature. The folded shape reverted to the original shape when it was heated to 50°C exhibiting the shape memory characteristics of the blended polyurethane.

#### EXAMPLE 4

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A polyurethane composition based on 1,3-bis(4-hydroxybutyl)tetramethyldisiloxane (BHTD) and MDI was prepared.

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BHTD (Silar Laboratories, 55.68 g) was added to molten (45°C) MDI (50.00g) and thoroughly mixed until a clear and homogenous solution was obtained. This required about 3 min of stirring. The viscous polymer was then poured onto a Teflon-coated metal tray and cured at 100°C for 4 h in an oven under nitrogen. The resulting polymer was clear and transparent. The cured polyurethane was compression moulded at 200°C to a 1 mm thick plaque. The materials exhibited a shore hardness of 75D, ultimate tensile strength of 60 MPa, and flexural modulus of 1795 MPa.

The onset of glass transition temperature was 30°C and the polyurethane remained rigid below 30°C and softened at body temperature (37°C).

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#### EXAMPLE 5

This example illustrates the preparation of a polyurethane using a low molecular weight siloxane macrodiol such that the polyurethane composition has a glass transition temperature close to the body temperature. The polyurethane was prepared by reacting 4,4'-methylenediphenyl diisocyanate (MDI, Orica), α, β-bis (6-hydroxyethoxypropyl)-polydimethylsiloxane (PDMS MW 595) and 1,4-cyclohexanedimethanol (Aldrich). PDMS with a molecular weight of 595 was obtained by distilling Shin-Etsu product X-22-160AS (Lot No. 803037) using a wiped-film evaporator.

PDMS was degassed at ambient temperature under vacuum (0.1 torr) for 4 h prior to polymerisation and CHDM (Aldrich) was melted at 60°C and degassed under vacuum (0.1 torr) for 1 h.

Degassed PDMS (5.94g) was added to molten (50°C) MDI (5.00g) in a polypropylene beaker and stirred rapidly until the solution turned clear followed by adding CHDM (1.44 g). After stirring the mixture for further 35 sec, the viscous polymer was poured onto a Teflon-coated pan and cured at 100°C for 6 h under nitrogen. Tensile properties

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were measured on a compression moulded sheet. DSC analysis was carried out to determine the glass transition temperature of the polyurethane. The polyurethane exhibited an ultimate tensile strength of 23.3 ± MPa, elongation at break of 97 ±8 % and a Young's modulus of 201±65. The DSC results showed the onset of glass transition to be 26°C, mid point at 34°C and end at 42°C. The polyurethane showed shape memory properties when tested using the procedure described in Example 3.

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#### EXAMPLE 6

This example illustrates the preparation of shape memory polyurethanes by blending commercial polyurethanes and a high modulus polyurethane with a glass transition temperature of about 100°C. PELLETHANE<sup>TM</sup> 2363-80A and CORETHANE<sup>TM</sup> AW 80 were used as examples of commercial polyurethanes.

The high modulus polyurethane was prepared using the following procedure. Molten (50°C) MDI (500.00 g) was weighed into a 2 L polypropylene beaker. The chain extenders BHTD (139.11 g) and CHDM (216.08 g) were weighed separately into two wet-tared polypropylene beakers. BHTD was added to MDI and stirred for about 45 seconds followed by molten (80°C) CHDM. Stirring was continued for another 20 to 25 sec and the viscous polymer was immediately stirred into a Teflon-coated tray. The tray containing the polymer was kept under nitrogen at ambient temperature for about 45 min and cured at 100°C for 4 h.

Two compositions were prepared by blending the high modulus polyurethane with CORETHANE<sup>TM</sup> and PELLETHANE<sup>TM</sup>, respectively. Composition 1 was prepared by dissolving 7.5 g of the high modulus polyurethane with 2.5 g of CORETHANE<sup>TM</sup> in 40 mL of dimethyl acetamide. The mixture was cast into a thin film by pouring the solution into a Petrie dish and evaporating the solvent in a nitrogen circulating oven at 70 °C for 48 h. Similarly Composition-2 was

prepared by dissolving 2.5 g of PELLETHANE<sup>TM</sup> and 7.5 g of the high modulus polyurethane in dimethylacetamide and casting a thin film of the composition.

The tensile properties and glass transition

temperature of the two compositions were determined and the
results are summarised in Table 1 below. The two
compositions showed shape memory properties when tested
using the procedure described in Example 3.

TABLE 1. Tensile properties and glass transition temperatures of the polyurethane compositions prepared in Example 6.

Composition Ele		UTS	TS YM Tg		Mid	Endpoint
	%	MPa	MPa	(°C)Onset	point	
Composition 1	13	41.5	648	39.8	43.2	46.7
Composition 2	13	28.0	280	44.6	48.3	52.0

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#### CLAIMS:

- 5 1. A shape memory polyurethane or polyurethane-urea polymer including a reaction product of:
  - (A) (a) silicon based macrodiol, silicon-based macrodiamine and/or polyether of the formula (I):

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$$A - [(CH_2)_m - O]_n - (CH_2)_m - A'$$

wherein

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A and A' are endcapping groups; m is an integer of 6 or more; and n is an integer of 1 or greater;

- (b) a diisocyanate; and
- (c) a chain extender; or
- (B) (b) a diisocyanate: and
- 20 (c) a chain extender,

said polymer having a glass transition temperature which enables the polymer to be formed into a first shape at a temperature higher than the glass transition temperature and maintained in said first shape when the polymer is cooled to a temperature lower than the glass transition temperature, said polymer then being capable of resuming its original shape on heating to a temperature higher than the glass transition temperature.

- 2. A shape memory polyurethane or polyurethane-urea
  30 polymer according to claim 1, wherein component (a) is a
  combination of at least two macrodiols, at least two
  macrodiamines or at least one macrodiol and at least one
  macrodiamine.
- 3. A shape memory polyurethane or polyurethane-urea
  5 polymer according to claim 1 or claim 2, wherein component
  (a) has greater than about 50% silicon-based macrodiol.
  - 4. A shape memory polyurethane or polyurethane-urea

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polymer according to any one of the preceding claims, wherein component (a) has greater than about 70% silicon based macrodiol.

- 5. A shape memory polyurethane or polyurethane-urea polymer according to any one of the preceding claims, wherein the molecular weight range of component (a) is about 300 to about 2000.
- 6. A shape memory polyurethane or polyurethane-urea polymer according to any one of the preceding claims, wherein the molecular weight range of component (a) is about 300 to about 700.
- 7. A shape memory polyurethane or polyurethane-urea polymer according to any one of the preceding claims, wherein the silicon-based macrodiol or macrodiamine is a polysilane, polysiloxane, amino-terminated polysiloxane or a silicon-based polycarbonate.
  - 8. A shape memory polyurethane or polyurethane-urea polymer according to claim 7, wherein the polysiloxane or amino-terminated polysiloxane is represented by the formula (II):

$$A-R_{5}-S_{i}-\begin{bmatrix} R_{1} \\ R_{7}-S_{i}-\\ R_{3} \end{bmatrix} \begin{bmatrix} R_{2} \\ R_{7}-S_{i}-\\ R_{4} \end{bmatrix} R_{6}-A'$$
(II)

wherein

A and A' are as defined in claim 1:

R<sup>1</sup>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub>, R<sub>5</sub> and R<sub>6</sub> are the same or

different and selected from hydrogen or an optionally substituted straight chain, branched or cyclic, saturated or unsaturated hydrocarbon radical;

R<sub>7</sub> is a divalent linking group or an optionally substituted straight chain, branched or cyclic, saturated or unsaturated hydrocarbon radical; and

p is an integer of 1 or greater.

9. A shape memory polyurethane or polyurethane-urea

polymer according to claim 8, wherein the divalent linking group for  $R_7$  is 0, S or NR wherein R is hydrogen or an optionally substituted straight chain, branched or cyclic, saturated or unsaturated hydrocarbon radical.

- 5 10. A shape memory polyurethane or polyurethane-urea polymer according to claim 8 or claim 9, wherein the polysiloxane is a polysiloxane macrodiol which is a polymer of the formula (II) wherein R and R' are hydroxy.
- 11. A shape memory polyurethane or polyurethane-urea
  10 polymer according to claim 10, wherein the polysiloxane
  macrodiol is represented by the formula (III):

$$HO-R_{5}-Si-\begin{bmatrix} R_{1} & R_{2} \\ R_{7}-Si- \\ R_{3} & R_{4} \end{bmatrix}_{R_{4}} R_{6}-OH$$
(III)

wherein

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R<sub>1</sub> to R<sub>6</sub> are as defined in claim 8.

- 12. A shape memory polyurethane or polyurethane-urea polymer according to claim 11, wherein the macrodiol is PDMS which is a compound of formula (III) wherein  $R_1$  to  $R_4$  are methyl and  $R_5$  and  $R_6$  are as defined in claim 11.
- 13. A shape memory polyurethane or polyurethane-urea 25 polymer according to claim 11, wherein R<sub>5</sub> and R<sub>6</sub> are the same or different and selected from propylene, butylene, pentylene, hexylene, ethoxypropyl (-CH<sub>2</sub>CH<sub>2</sub>OCH<sub>2</sub>CH<sub>2</sub>-), propoxypropyl and butoxypropyl.
- 14. A shape memory polyurethane or polyurethane-urea polymer according to any one of claims 10 to 13, wherein the molecular weight range of the polysiloxane macrodiol is about 200 to about 6000.
  - 15. A shape memory polyurethane or polyurethane-urea polymer according to any one of claims 10 to 14, wherein the molecular weight range of the polysiloxane macrodiol is about 500 to about 2000.
    - 16. A shape memory polyurethane or polyurethane-urea

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polymer according to claim 8 or claim 9, wherein the aminoterminated polysiloxane is a polysiloxane macrodiamine which is a polymer of the formula (II) wherein A is NH2.

- 17. A shape memory polyurethane or polyurethane-urea polymer according to claim 16, wherein the polysiloxane macrodiamine is amino-terminated PDMS.
- 18. A shape memory polyurethane or polyurethane-urea polymer according to any one of claims 7 to 17, wherein the silicon-based polycarbonate has the formula (IV):

$$= \begin{bmatrix} R_2 \\ R_7 - Si \\ R_4 \end{bmatrix}_m - R_6 - \begin{bmatrix} O \\ O - C - O - R_9 \\ 0 \end{bmatrix}_y - \begin{bmatrix} O \\ R_1 \\ O - C - O - R_9 \\ 0 \end{bmatrix}_z - \begin{bmatrix} O \\ R_1 \\ R_3 \end{bmatrix}_m \begin{bmatrix} R_2 \\ R_7 - Si \\ R_4 \end{bmatrix}_m - R_6 - A^2$$

(IV)

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 $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$ ,  $R_5$ ,  $R_6$  and  $R_7$  are as defined in formula (I) in claim 1;

R<sub>8</sub> and R<sub>9</sub> are same or different and selected from hydrogen or an optionally substituted straight chain, branched or cyclic, saturated or unsaturated hydrocarbon radical;

> A and A'are as defined in formula (I) in claim 1; m, y and z are integers of 0 or more; and x is an integer of 0 or more.

19. A shape memory polyurethane or polyurethane-urea 35 polymer according to claim 18, wherein z is an integer of 0 to about 50, x is an integer of 1 to about 50, m is an

integer of 0 to about 20 and y is an integer of 0 to about 10.

20. A shape memory polyurethane or polyurethane-urea polymer according to claim 18 or claim 19, wherein the silicon-based polycarbonate is a compound of the formula (IV) wherein the endcapping group is a hydroxy which is a polycarbonate macrodiol of the formula (IVa):

$$\begin{array}{c}
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H-O-R_{5}-\stackrel{\downarrow}{Si-} \begin{bmatrix} R_{2} \\ R_{7}-\stackrel{\downarrow}{Si} \\ R_{4} \end{bmatrix}_{m} -R_{6}-O-\stackrel{\downarrow}{C} \begin{bmatrix} O-R_{5}-\stackrel{\downarrow}{Si-} \begin{bmatrix} R_{2} \\ R_{7}-\stackrel{\downarrow}{Si} \\ R_{3} \end{bmatrix} -R_{6}-O-\stackrel{\downarrow}{C} \end{bmatrix}_{x} -R_{6}-O-\stackrel{\downarrow}{C} \begin{bmatrix} O-R_{8} \\ O-C-O-R_{9} \end{bmatrix}_{y} - \begin{bmatrix} O-R_{5}-\stackrel{\downarrow}{Si-} \begin{bmatrix} R_{2} \\ R_{7}-\stackrel{\downarrow}{Si} \\ R_{3} \end{bmatrix} -R_{6}-O-\stackrel{\downarrow}{C} \end{bmatrix}_{x} -R_{6}-O-\stackrel{\downarrow}{C} \\
0 -C-O-R_{9} \end{bmatrix}_{z} -O-\stackrel{\downarrow}{C}-O-R_{5}-\stackrel{\downarrow}{Si-} \begin{bmatrix} R_{2} \\ R_{7}-\stackrel{\downarrow}{Si} \\ R_{3} \end{bmatrix} -R_{6}-O-\stackrel{\downarrow}{C} \\
0 -C-O-\stackrel{\downarrow}{C} \\$$

wherein

 $R_1$  to  $R_9,\ m,\ y,\ x$  and z are as defined in formula (IV) in claim 18.

- 21. A shape memory polyurethane or polyurethane-urea
  25 polymer according to claim 20, wherein the polycarbonate macrodiol is a compound of the formula (IVa) wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub> and R<sub>4</sub> are methyl, R<sub>8</sub> is ethyl, R<sub>9</sub> is hexyl, R<sub>5</sub> and R<sub>6</sub> are propyl or butyl and R<sub>7</sub> is 0 or -CH<sub>2</sub>-CH<sub>2</sub>-.
- 22. A shape memory polyurethane or polyurethane-urea
  30 polymer according to claim 21 wherein R<sub>5</sub> and R<sub>6</sub> are propyl
  when R<sub>7</sub> is 0 or R<sub>5</sub> and R<sub>6</sub> are butyl when R<sub>7</sub> is -CH<sub>2</sub>-CH<sub>2</sub>-.
  23. A shape memory polyurethane or polyurethane-urea
  polymer according to any one of claims 20 to 22, wherein
  the molecular weight range of the polycarbonate macrodiol

is about 400 to about 5000.

24. A shape memory polyurethane or polyurethane-urea polymer according to any one of claims 20 to 23, wherein

the molecular weight range of the polycarbonate macrodiol is about 400 to about 2000.

25. A shape memory polyurethane or polyurethane-urea polymer according to any one of the preceding claims,
5 wherein the polyether is a polyether macrodiol represented by the formula (V):

$$HO-[(CH_2)_m-O]_n-H$$

10 wherein

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m is as defined in formula (I) in claim 1; and n is as defined in formula (I) in claim 1.

- 26. A shape memory polyurethane or polyurethane-urea polymer according to claim 25, wherein the polyether

  15 macrodiol is poly(tetramethylene oxide) (PTMO), poly(hexamethylene oxide) (PHMO), poly(hexamethylene oxide) (POMO) or poly(decamethylene oxide) (PDMO).
- 27. A shape memory polyurethane or polyurethane-urea polymer according to claim 25 to claim 27, wherein the molecular weight range of the polyether macrodiol is about 300 to about 2000.
  - 28. A shape memory polyurethane or polyurethane-urea polymer according to any one of claims 25 to 27, wherein the molecular weight range of the polyether macrodiol is about 300 to 700.
  - 29. A shape memory polyurethane or polyurethane-urea polymer according to any one of claims 12 to 28, wherein component (a) is a combination of PDMS or amino-terminated PDMS with another polymer falling within the scope of component (a).
  - 30. A shape memory polyurethane or polyurethane-urea polymer according to claim 29, wherein said another polymer is a polyether of the formula (I) or a silicon based polycarbonate.
  - 31. A shape memory polyurethane or polyurethane-urea polymer according to claim 30, wherein the polyether of the

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formula (I) is PHMO.

- A shape memory polyurethane or polyurethane-urea polymer according to claim 30 or claim 31, wherein the silicon-based polycarbonate is siloxy carbonate.
- A shape memory polyurethane or polyurethane-urea polymer according to any one of the preceding claims, wherein the diisocyanate is an aliphatic or aromatic diisocyanate.
- A shape memory polyurethane or polyurethane-urea polymer according to any of the preceding claims, wherein the diisocyanate is 4,4'-diphenylmethane diisocyanate (MDI), methylene biscyclohexyl diisocyanate (H12MDI), pphenylene diisocyanate (p-PDI), trans-cyclohexane-1,4diisocyanate (CHDI), 1,6-diisocyanatohexane (DICH), 1,5-
- diisocyanatonaphthalene (NDI), paratetramethylxylenediisocyanate (p-TMXDI), metatetramethylxylene diisocyanate (m-TMXDI), 2,4-toluene diisocyanate (2,4-TDI) isomers or mixtures thereof or isophorone diisocyanate (IPDI).
- 20 35. A shape memory polyurethane or polyurethane-urea polymer according to any one of the preceding claims, wherein the diisocyanate is MDI.
  - A shape memory polyurethane or polyurethane-urea polymer according to any one of the preceding claims,
- 25 wherein the chain extender is a diol or diamine chain extender.
  - A shape memory polyurethane or polyurethane-urea polymer according to claim 36, wherein the diol chain extender is 1,4-butanediol, 1,6-hexanediol, 1,8-octanediol,
- 1,9-nonanediol, 1,10-decanediol, 1,12-dodecanediol, 1,4cyclohexanediol, 1,4-cyclohexanedimethanol, p-xyleneglycol, 1,3-bis(4-hydroxybutyl)tetramethyldisiloxane, 1,3-bis(6hydroxyethoxypropyl)tetramethyldisiloxane or 1,4-bis(2hydroxyethoxy) benzene.
- A shape memory polyurethane or polyurethane-urea polymer according to claim 36, wherein the diamine chain extender is 1,2-ethylenediamine, 1,3-propanediamine, 1,4-

butanediamine, 1,3-bis(3-aminopropyl)tetramethyldisiloxane, 1,3-bis(4-aminobutyl)tetramethyldisiloxane or 1,6-hexanediamine.

39. A shape memory polyurethane or polyurethane-urea
5 polymer according to any one of claims 36 to 38, wherein
the chain extender is a silicon-containing chain extender.
40. A shape memory polyurethane or polyurethane-urea
polymer according to claim 39, wherein the siliconcontaining chain extender includes a silicon-containing
10 diol of the formula (VI):

HO-
$$R_5$$
— $\begin{bmatrix} R_1 \\ R_2 \\ R_7$ — $\begin{bmatrix} R_2 \\ R_7$ — $\begin{bmatrix} R_2 \\ R_4 \end{bmatrix} \end{bmatrix}$ — $R_6$ —OH

wherein

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 $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$ ,  $R_5$ ,  $R_6$  and  $R_7$  are as defined in formula (II) in claim 8; and

q is 0 or greater.

- 41. A shape memory polyurethane or polyurethane-urea polymer according to any one of the preceding claims, wherein component (a) polymer forms the soft segment of the polyurethane or polyurethane-urea polymer.
- 42. A shape memory polyurethane or polyurethane-urea polymer according to any one of the preceding claims, wherein components (b) and (c) of the polymer form the hard segment of the polyurethane or polyurethane-urea polymer.
  - 43. A shape memory polyurethane or polyurethane-urea polymer according to claim 42, wherein the amount of hard segment in the polymer is about 30 to 100 wt%.
  - 44. A shape memory polyurethane or polyurethane-urea polymer according to claim 42 or 43, wherein the amount of hard segment in the polymer about 50 to about 80 wt%.
- 45. A shape memory polyurethane or polyurethane-urea polymer according to any one of claims 42 to 44, wherein the amount of hard segment in the polymer is about 60 to

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about 70 wt%.

- 46. A shape memory polyurethane or polyurethane-urea polymer according to any one of the preceding claims, wherein the shore hardness of the polymer below the glass transition temperature is in the range of about 82D to about 50D, while the hardness above the glass transition temperature is in the range of about 20D to about 30D.
- 47. A shape memory polyurethane or polyurethane-urea polymer according to any one of the preceding claims, wherein the glass transition temperature is in the range of about 20°C to about 100°C.
- 48. A shape memory polyurethane or polyurethane-urea polymer according to any one of the preceding claims, wherein the glass transition temperature is in the range of about 20 to about 60°C.
- 49. A shape memory composition which includes a blend of two or more of the shape memory polyurethane or polyurethane-urea polymers defined in any one of the preceding claims or at least one shape memory polyurethane or polyurethane-urea polymer defined in any one of the preceding claims in combination with another material.
- A shape memory composition according to claim 49, wherein the other material is a polymeric or a nonpolymeric material.
- A shape memory composition according to claim 50, 25 wherein the polymeric material is a conventional polyurethane, shape memory polyurethane, polyolefin, polyamide or a liquid crystalline polymer.
- A shape memory composition according to any one of claims 49 to 51, wherein each of the polymers forming the shape memory composition have different glass transition temperatures and/or different amounts of hard segment component.
- A shape memory composition according to claim 52, which includes a first polymer with a low glass transition temperature of below about ambient temperature and a second polymer with a glass transition temperature above the

ambient temperature.

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54. A shape memory composition according to claim 52 or claim 53, wherein the two polymers can be blended in proportions such that the final blend will have a glass transition temperature in the range of about 20°C to about 60°C.

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- 55. A shape memory composition according to claim 52, which includes a first polymer having a high percentage of hard segment component of above about 70 wt% and a second polymer having a lower percentage of hard segment of about 30 to about 60wt%.
- 56. A shape memory composition according to claim 55, wherein the composition includes a combination of an elastomeric and a non-elastomeric polyurethane or polyurethane-urea polymer.
- 57. A process for preparing a shape memory polymer as defined in any one of claims 1 to 48 which includes the steps of:
- (i) mixing component (a) and the chain extender 20 (c); and
  - (ii) reacting the mixture with the diisocyanate (b).
  - 58. A process according to claim 57, wherein step (i) is performed at a temperature in the range of about  $45^{\circ}$ C to about  $100^{\circ}$ C.
  - 59. A process according to claim 57 or claim 58, wherein step (i) occurs in the presence of a catalyst.
  - 60. A process for preparing a shape memory polymer as defined in any one of claims 1 to 48 which includes the steps of:
  - (i) reacting component (a) with a diisocyanate(b) to form a prepolymer; and
  - (ii) reacting the prepolymer with the chain extender (c).
- 35 61. A material having improved mechanical properties, clarity, processability, biostability and/or degradation resistance which includes the shape memory polymer as

defined in any one of claims 1 to 48 and/or the composition as defined in any one of claims 49 to 56.

- 62. A material according to claim 61, wherein the improved mechanical properties are tensile strength, tear strength,
- flex fatigue resistance, abrasion resistance, Durometer hardness, flexural modulus and/or related measures of flexibility or elasticity.
  - 63. A material according to claim 61 or 62, wherein the improved resistance to degradation is resistance to free radical, oxidative, enzymatic and/or hydrolytic processes and/or to degradation when implanted as a biomaterial.
  - 64. A material according to any one of claims 61 to 63, wherein the improved processability is ease of processing by casting and/or thermal means.
- 15 65. A material according to any one of claims 61 to 64, which is a biostable material.
  - 66. A material according to any one of claims 61 to 65, which is a degradation resistant material.
  - 67. A material according to any one of claims 61 to 66, which is an *in vivo* degradation resistant or biostable material.
    - 68. A material according to any one of claims 61 to 67, which is a biomaterial.
- 69. Use of the shape memory polymer as defined in any one of claims 1 to 48 and/or composition as defined in any one of claims 49 to 56 as a material having improved mechanical properties, clarity, processability, biostability and/or degradation resistance.
- 70. The shape memory polymer as defined in any one of claims 1 to 48 and/or composition as defined in any one of claims 49 to 56 when used as a material having improved mechanical properties, clarity, processability, biostability and for degradation resistance.
- 71. A device or article which is composed wholly or partly of the shape memory polymer as defined in any one of claims 1 to 48 and/or composition as defined in any one of claims 49 to 56.

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- 72. A device or article according to claim 71, which is a medical device, article or implant.
- 73. A device or article according to claim 72, which is a stylet; bone suture anchor; vascular, oesophageal or bilial stent; cochlear implant; reconstructive facial surgery; controlled drug release device; component in key-hole surgery; biosensor; membrane for cell encapsulation; medical guidewire; medical guidepin; cannularization; pacemaker, defibrillator or neurostimulator and their
- respective electrode leads; ventricular assist device; orthopaedic joint or parts thereof; intraoccular lens; urological device; stent/graft device; device joining/extending/repair sleeves; heart valve; vein graft; vascular access port; vascular shunt; blood purification
- device; cast for a broken limb; vein valve, angioplasty, electrophysiology or cardiac output catheter; or tools for insertion of medical devices, infusion and flow control devices.
- 74. A device or article according to claim 71, which is a toy or component thereof, shape memory film, pipe coupling, electrical connector, zero-insertion force connector, robotic, aerospace actuator, dynamic display, flow control device, sporting goods and components thereof, body-conforming device, temperature control device, safety release device or heat shrink insulation.
  - 75. Use of the shape memory polymer as defined in any one of claims 1 to 48 and/or composition as defined in any one of claims 49 to 55 in the manufacture of a device or article.
- 76. A shape memory polymer as defined in any one of claims 1 to 48 and/or a composition as defined in any one of claims 48 to 55 when used in manufacture of a device or article.

## INTERNATIONAL SEARCH REPORT

International application No.

PCT/AU00/00863

A.	CLASSIFICATION OF SUBJECT MATTER			
Int. Cl. 7:	C08G 18/61, 18/48, A61L 27/00, 29/00, 31/0	00		
According to	International Patent Classification (IPC) or to bot	h national classification and IPC		
В.	FIELDS SEARCHED			
Minimum docu C08G 18/61	umentation searched (classification system followed by , 18/48	classification symbols)		
Documentation AU: IPC as a	searched other than minimum documentation to the exabove	ctent that such documents are included in t	the fields searched	
Electronic data WPAT & JA	base consulted during the international search (name of	f data base and, where practicable, search	terms used)	
C.	DOCUMENTS CONSIDERED TO BE RELEVAN	Т		
Category*	Citation of document, with indication, where ap	propriate, of the relevant passages	Relevant to claim No.	
х	AU 41924/97 (CARDIAC CRC NOMINEES PTY LTD) 17 April 1998 Page 6 line 23 - page 13 line 18, Examples 1-20 and claims 1-48  1-76		1-76	
x	US 5911737A (LEE et al.) 15 June 1999 Column 3 lines 10-20, column 3 line 51 - column 4 line 13,		1-76	
x	US 5139832A (HAYASHI et al.) 18 August Column 2 line 25 - column 3 line 27, Examp		1-76	
x	Further documents are listed in the continuation	on of Box C X See patent fam	ily annex	
"A" document to come the interest of the inter	not considered to be of particular relevance  "E" earlier application or patent but published on or after the international filing date  "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)  "O" document referring to an oral disclosure, use, exhibition or other means  "I" understand the principle or theory underlying the invention cannot document of particular relevance; the claimed invention cannot be considered novel or cannot be considered novel or cannot be considered to involve an inventive step when the document is combined with one or more other such document is combined with one or more other such documents, such combination being obvious to a person skilled in the art			
1	Date of the actual completion of the international search  2 August 2000  Date of mailing of the international search report  AUG 2000			
	lame and mailing address of the ISA/AU Authorized officer			
AUSTRALIAN PO BOX 200, V E-mail address:	PATENT OFFICE WODEN ACT 2606, AUSTRALIA : pct@ipaustralia.gov.au (02) 6285 3929	ALBERT S. J. YONG Telephone No : (02) 6283 2160		

## INTERNATIONAL SEARCH REPORT

International application No.

PCT/AU00/00863

C (Continua	tion). DOCUMENTS CONSIDERED TO BE RELEVANT	
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
х	US 5049591A (HAYASHI et al.) 17 September 1991 Column 2 lines 22-33, Table 1	1-76
x	US 5430121A (PUDLEINER et al.) 4 July 1995 Column 4 lines 3-53, column 8 lines 30-36	1-76
x	US 5393858A (MEIJS et al.) 28 February 1995 Column 2 line 26 - column 3 line 41, Example 2	1-76
x	US 4786657A (HAMMAR et al.) 22 November 1988 Examples 12, 19, 20	1-76
	Derwent Accession No. 92-344628/42, Class P34, JP 4-248826A (TOYOBO KK) 4 September 1992	
х	See Abstract  Derwent Accession No. 95-325584/42, Class A25, JP 7-224138A (SANYO CHEM) 22	1-76
x	August 1995 See Abstract	1-76
· <b>x</b>	Derwent Accession No. 88-246748/35, Class A25, JP 63-179916A (DAINIPPON INK CHEM) 23 July 1988 See Abstract	1-76
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# INTERNATIONAL SEARCH REPORT Information on patent family members

International application No. PCT/AU00/00863

This Annex lists the known "A" publication level patent family members relating to the patent documents cited in the above-mentioned international search report. The Australian Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

Patent Document Cited in Search Report		rch	Patent Family Member					
AU	41924/97	wo	98/13405	EP	938512			
US	5911737	AU	63432/98	wo	98/37816			
US	5139832	CA	1321461	EP	363919	JP	2106324	
US	5049591	CA	1319238	EP	361418	JP	2092912	
US	5430121	CA	2111925	DE	4243799	EP		
US	5393858	AU	80065/91	EP	536223	wo	92/00338	
US	4786657	AU	17306/88	CA	1333948	EP	298611	
		JP	1033114					
JP	4-248826	NONE				•		
JP	7-224138	NONE						
JР	63-179916	NONE						
						E	ND OF ANN	